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Cyclohexane contains only single bonds and is representative of the simplest type of cyclic hydrocarbons. A ring structure may possess double bonds, as in the following portrayal of the well-known hydrocarbon benzene, which has the composition C₆H₆. See Figure 8. Figure 8. Benzene.

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[MOBI] Chemistry Study Guide Hydrocarbons Answer Study Guide Key Concepts 22.1 Hydrocarbons

- Because carbon has four valence electrons, carbon atoms always form four covalent bonds.
 - The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches.
 - Molecules of hydrocarbons, such as alkanes, are nonpolar molecules.
- 22.2 Unsaturated Hydrocarbons

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Chemists called the hydrocarbons that reacted with bromine unsaturated hydrocarbons in the same sense that an unsaturated aqueous solution can dissolve more solute. Hydrocarbons that did not react with bromine were said to be saturated. Present-day chemists can now explain the experimental results obtained 170 years ago.

Chapter 21: Hydrocarbons

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